

Research Journal of Recent Sciences Vol. **12(3)**, 16-20, October (**2023**)

Extraction of natural dye from *Rhus parviflora* (Tung) and evaluation of colour fastness properties of weaved wool herbal dyed fabrics using mixture of bio and chemical mordants

Naveen Kumar¹, Shyam Vir Singh^{2*} and M.C. Purohit³

¹Department of Chemical Science, KLDAV PG College, Roorkee, Haridwar, UK, India ²Department of Chemical Science, S.G.R.R. (Post Graduate) College, Pathribagh, Dehradun, UK, India ³Department of Chemical Science, H.N.B. Garhwal University (A Central University), Campus Pauri, Pauri Garhwal, UK, India shyamveer91084@rediffmail.com

> **Available online at: www.isca.in, www.isca.me** Received 29th June 2023, revised 23rd August 2023, accepted 30th September 2023

Abstract

Present research is an attempt to assess the fastness grades of colourant on weaved woolen fabrics which were dyed by herbal dye extracted by the leaves of Rhus parviflora has been investigated with a mixture of botanical and chemical colour fixing agents like lime extract + $CuSO_4$, lime extract + $K_2Cr_2O_7$, lime extract + $FeSO_4$ and lime extract + $SnCl_2$ with the ratio 3:1, 1:2 and 1:3 respectively. Here dyeing method was followed with some colour fixing agent methods like prechromiccolour fixing agent, metachromic colour fixing agent and after chromic-colour fixing agent. Fastness tests of dyed fabrics are also observed under this study. A wide range of colour shades were observed with altering colour fixing agent ratios and their mixture. The fastness grades of the dyed fabrics to wash, rub, light and perspiration have also been examined as fair to excellent fastness grades and this observation can also be helpful to textile industries.

Keywords: Herbal dye, Wool fabrics, *Rhus parviflora*, Bio and chemical mordants or colour fixing agents, Colour fastness properties.

Introduction

In ancient times, most of the people were depend on the natural sources for textile dyeing, such as flowers fruits, leaves, seeds bark, the root of plants etc. Plant colourants have been used for dyeing textiles like wool, silk, cotton and various bio-fibers across the globe. Due to its eco-friendly environment nature and extraordinary colour diversity, it creates a non-polluted environment and these colourants have been used for colouring textiles as wool and other bio-fibers across the globe by the various human societies^{1,2} and their fastness grades vary with nature of soil and the weather³⁻⁷. The colour components responsible for dyeing were isolated and their chemical constituents were established based on chemical and spectroscopic investigations. Generally, colour fixing agents can be botanical (as an extract of leaves, flowers, stem and root etc.) and chemical (as a metal salts which can be responsible for creating chemistry between the fabrics and herbal dye). Lodhspecies is actively giving the herbal dyes properties and this species give us an excellent colour grades w.r.t. grey scale. In this research, the aqueous extract of tung is attempted to colouring of woolen fabric sat some refined dyeing stages which followed with the mixture of botanical and chemical colour fixing agents⁸⁻¹¹ and these dyed fabrics were examined with respect to fastness as wash, rub, perspiration and light¹²⁻¹⁵. There are many natural resources are responsible to dyeing for the industries. However, the common drawbacks of herbal dyes are their non-recreatable nature, multiform shades, moderate

colour grades and lack of sufficient scientific information on the chemistry for colouring with respective finishing conditions. Review of literature about finishing of textiles with herbal dye revealed lots of information to the application of herbal dyes using bio and chemical colour fixing agents with different colour characteristics¹⁶ followed by different colour fixing agent methods like prechromic-colour fixing agent, metachromic colour fixing agent.

Status of eco-friendly natural dyes and dye bearing plants in India: According to literature, the main use of nature friendly has been show versatility to exploring the finishing of textiles or polishing of medicinal drugs and colouring of beauty products etc. Due to its non-toxic nature, herbal dyes are also responsible to colouring of food products. Literature revealed that In India, appearance of more than 450 plants can bear the dyes¹⁷ but unfortunately there are not sufficient data resources to know about applicability or availability of natural dyes towards extracting and dyeing techniques as well as too much information among the society to utilization of natural dyes over synthetic dyes.

Materials and methods

Rhus parviflora (Tung) is commonly known as Tung or Tunn and found in various places of the Garhwal region of Uttarakhand. For this research, leaves of lodh plant were taken for dyeing which was followed by aqueous extraction and bleached plain weaved woolen fabrics purchased from the district Chamoli, Uttarakhand. Chemical metal salts (analytical reagents (AR) grade) were used as chemicalcolour fixing agents such as $K_2Cr_2O_7$, CuSO₄, SnCl₂ and FeSO₄. Some laboratory reagents as acetic acid, common salt and sodium carbonate were also used as same and a bio-colour fixing agents 'lime extract' was used for this research respectively^{8,10}.

By using the aqueous extracted herbal dye for dyeing from the leaves of *Rhus parviflora*, various colour shades were observed by using the different metal salts as chemical colour fixing agents and these colour shades of dyed fabrics were intervene with lime extract when it mixed with a standard vol. of distilled water and heating at 80° C for 30 min was followed by three colour fixing agent conditions and this solution was cooled and filtered, this filtrate was responsible for different conditions of colour fixing agent. Now took the shaded dried leaves of *Rhus parviflora* with a known quantity and soaked in lukewarm water entire night and then obtained extract was further boiling with distilled water and cooled, filtered and then finally dyeing to weave woolen fabrics.

Fabrics dyeing were occurred with some optimized conditions like extraction time for dye (60 min), M-L ratio as 1:20 and time for dveing (50 min.). The herbal and chemical colour fixing agent mixtures viz. lime extract: CuSO₄, lime extract: K₂Cr₂O₇, lime extract: FeSO₄, lime extract: SnCl₂were used with the ratio 3:1, 1:2 and 1:3 respectively. Total quantity of each two colour fixing agent mixture was 5% on the weight of the woolen fabric i.e. 5g of the colour fixing agent /100 g of the fabric. These four colour fixing agent mixtures was under go with three different ratios as 3:1, 1:2 and 1:3 for three colour fixing agent conditions such as prechromic colour fixing agent, met chromic colour fixing agent and after chromic colour fixing agent for dyeing. Hence after fabrics dveing, the solution was cooled then dved fabrics were taken out from the dye bath, wash with distilled water to minimize the excess dye colourant and were dried in shade.

Measurement of fastness grades towards light, washing, rubbing (dry and wet) and perspiration were occurred with the help of ultra-violet light in a Shirley MBTF Microsal fade-O-meter as per ISO: 2454-1984 and fade nature of fabrics were measured by comparing with blue wool standards (BS1006: BOI: 1978), Sasmira launder-O-meter followed by IS-3 wash fastness method as per ISO-05-A02 (loss of shade depth) and ISO-105-AO3 (proportion of staining) and staining was measured with Macbeth 2020 plus computer-ACM system, a manually operatable crock-meter and grey scale as per ISO-105-AO3 (proportion of staining) as per IS: 766-1984 method and perspiro-meter under load of 4.5 kgs (10 lbs) was responsible for measuring the perspiration fastness grades in the acidic and alkaline media with M-L ratio 1:50 for 30 min at 25°C. At last, specimen were taken out and dried in the air with the temperature not more than 60°C and colour staining and change was occurred using grey scale^{8,10}.

Results and discussion

Colour fixing agent mixture – Lime extract: stannous chloride: In this case, evaluated fastness to light was remain same for all the ratios of color fixing agent mixtures while observed fastness was excellent to good (4-5 to 4) for washing under metachromic and after chromic colour fixing agent methods. Change in colour for dry and wet rubbing fastness condition, it was excellent whereas under staining, was fluctuated from non-staining to negligible (5 to 4-5) in dry case. Except prechromic colour fixing agent condition, it was found that perspiration fastness grades fluctuated from excellent to good (4-5 to 4) for all in acidic and alkaline media under no colour staining in both media (Table-1).

Colour fixing agent mixture – Lime extract: copper sulphate: Under this condition, evaluated fastness grades were observed as fairly good (4) to light and fair (4 to 5) to wash under prechromic condition but it was excellent to good (4 - 5) for both metachromic and after chromic respectively. Fastness to rub with dry and wet condition, it was excellent for all cases. There is no observation for colour staining but change in colour to dry and wet was observed as an excellent (5) for all cases. Except for 1:3 colour fixing agent ratio in prechromic colour fixing agent condition, there fastness grades were fluctuated from good to excellent (4-5) for perspiration fastness in both media but there was no staining of colour for all observed samples (Table-2).

Cooler fixing agent mixture – Lime extract: potassium dichromate: In this case, observed fastness was fairly good (4) to light for all cases and like fastness to light, wash fastness was fluctuated from fair (3-4) to fairly good (4) for all conditions except 1:3 colour fixing agent mixture under prechromic condition where it was fair. For fastness grade to rubbing, change in colour to dry and wet was observed as an excellent (5) for all cases. Except 1:3 colour fixing agent ratio under prechromic condition where it was good (4-5) and there was no colour staining observed in both the media (Table-3).

Colour fixing agentmixture – Lime extract: ferrous sulphate: For this condition, it was observed that fastness grade to light was recorded as fairly good (4) in all cases whereas washing fastness grades were fluctuated from fair to good (3-4 to 4-5) almost for all. Under rubbing condition, it was observed as almost an excellent in all cases while fastness grades were almost excellent in the absence of staining (5) to perspiration in the respective media except after chromic method where it was observed as (4) Table-4.

Aqueous extracted eco-friendly herbal dye from leaves of *Rhus* parviflora was pale yellow in colour. By using different colour fixing agent condition with some chemical colour fixing agents like $K_2Cr_2O_7$, $CuSO_4$, $SnCl_2$ and $FeSO_4$, we observed beautiful colour shades than extracted herbal dye without any colour fixing agent as mentioned in Table-5.

Table-1: Colour fastness grades of dyed weaved woolen fabrics with *Rhus parviflora* dye at some refined conditions using herbal and chemical colour fixing agent mixture as LJ: SC under three colour fixing agent conditions.

Colour fixing agent method	Colour fixing agent	Fastness grades to light	Fastness grades to washing		Fastness grades troubling			Fastness grades to Perspiration				
method	proportions		CC	CS	Dry		Wet	Acidic Alka		caline		
					CC	CS	CC	CS	CC	CS	CC	
Prechromic	3:1	4	2-3	5	4-5	5	4	5	3	5	3	
colour fixing	1:2	4	2-3	5	5	4-5	5	5	4-5	5	4	
agent	1:3	4	2-3	5	5	4-5	5	5	4-5	5	4	
Metachromic	3:1	4	4-5	5	5	5	5	5	4	5	4	
colour fixing	1:2	4	4	5	5	5	5	5	4	5	4	
agent	1:3	4	4	5	5	5	5	5	4	5	4	
Afterchromic	3:1	4	4	5	5	4-5	5	5	4	5	4-5	
colour fixing	1:2	4	4-5	5	5	4-5	5	5	4-5	5	4-5	
agent	1:3	4	4-5	5	5	4-5	5	5	4-5	5	4-5	

LJ: SC- Lime extract: SnCl₂, CC -Change in colour, CS - Staining of colour.

Table-2: Colour fastness grades of dyed weaved woolen fabrics with *Rhus parviflora* dye at some refined conditions using herbal and chemical colour fixing agent mixture as LJ: CS under three colour fixing agent conditions.

Colour fixing agent	Colour fixing agent	Fastness grades to light	Fastness grades to washing		Fastness grades to rubbing			Fastness grades to Perspiration				
method	proportions	_			Dry		Wet	Acidic		Alkaline		
			CC	CS	CC	CS	CC	CS	CC	CS	CC	
Prechromic	3:1	4	3-4	5	5	5	5	5	4	5	4	
colour fixing	1:2	4	4	5	5	5	5	5	3	5	4	
agent	1:3	4	4	5	5	5	5	5	3	5	4	
Metachromic	3:1	4	4	5	5	5	5	5	3	5	4	
colour fixing	1:2	4	4	5	5	4-5	5	5	3	5	4	
agent	1:3	4	4	4-5	5	4-5	5	5	4	5	4	
Afterchromic	3:1	4	4	4-5	5	4-5	5	5	4	5	4-5	
colour fixing	1:2	4	4	4-5	5	4-5	5	5	4	5	4-5	
agent	1:3	3-4	4	4-5	5	4-5	5	5	4	5	4-5	

LJ: CS- Lime extract: CuSO₄, CC - Change in colour, CS - Staining of colour.

Table-3: Colour fastness grades of dyed weaved woolen fabrics with <i>Rhus parviflora</i> dye at some refined conditions using herbal
and chemical colour fixing agent mixture as LJ: PD under three colour fixing agent conditions.

Colour fixing	Colour fixing agent	Fastness	Fastness grades		Fastness grades to rubbing			Fastness grades to Perspiration			
agent	proportions	grades to	to was	to washing		Dry		Acidic		Alkaline	
method		light	CC	CS	CC	CS	CC	CS	CC	CS	CC
Prechromic	3:1	4	3	5	5	4-5	3-4	5	4-5	5	3
colour fixing	1:2	3-4	2-3	5	5	4-5	5	5	3-4	5	4
agent	1:3	4	2-3	5	5	4-5	5	5	4-5	5	4
Metachromic	3:1	4	2-3	5	5	5	5	5	4	5	4
colour fixing	1:2	4	4	5	5	5	5	5	4	5	4
agent	1:3	4	4	5	5	5	5	5	4	5	4
Afterchromic	3:1	3-4	3	5	5	5	5	5	4	5	4-5
colour fixing	1:2	3-4	4-5	5	5	5	5	5	4-5	5	4
agent	1:3	3-4	4-5	5	4-5	5	5	5	4-5	4-5	4-5

LJ: PD- Lime extract: K₂Cr₂O₇, CC -Change in colour, CS - Staining of colour.

Table-4: Colour fastness grades of dyed weaved woolen fabrics with *Rhus parviflora* dye at some refined conditions using herbal and chemical colour fixing agent mixture as LJ: FS under three colour fixing agent conditions.

Colour	Colour		Fastness grades		Fastness grades to rubbing			Fastness grades to Perspiration			
fixing agent	fixing agent fixing agent	grades to	to washing		Dry		Wet	Acidic		Alkaline	
method	proportions	light	CC	CS	CC	CS	CC	CS	CC	CS	CC
Prechromic	3:1	4	3-4	4-5	5	5	5	5	5	5	4
colour	1:2	4	4	4-5	5	5	5	4-5	5	5	4-5
fixing agent	1:3	4	3-4	4-5	5	5	5	5	5	5	5
Metachromi	3:1	4	3-4	4-5	5	4-5	5	5	5	5	5
с	1:2	4	3-4	4-5	5	5	5	5	5	5	5
colour fixing agent	1:3	4	3-4	4-5	5	5	5	5	5	5	5
Afterchromi	3:1	4	3-4	4-5	5	5	5	5	4	5	4-5
c colour	1:2	4	3-4	4-5	5	5	5	5	4	5	4-5
fixing agent	1:3	4	4	4-5	5	5	5	5	4	5	4-5

LJ: FS- Lime extract: FeSO₄, CC - Change in colour, CS - Staining of colour.

Table-5: Observed colour shades of dyed weaved woolen fabrics using mixture of chemical and bio-colour fixing agents (with and without colour fixing agent condition).

Colour fixing agent conditions Used	Dyeing (without colour fixing agents)	Dyeing with the mixture of chemical and bio- colour fixing agents (lime extract)
Prechromic Colour fixing agent		A States
Metachromic Colour fixing agent	R	Time Time Time Time Time Time Time Time
Afterchromic Colour fixing agent		A JT PAR A DE MILES A DE MIL

Conclusion

Under this section, it was observed from this study that *Rhus parviflora* plant is responsible to give best dyeing properties towards weaved wool fabrics or fibers including fastness grades expose to light, wash, rub and perspiration in which fairly good fastness to light, excellent grades to rubbing and washing excluding prechromic condition for both cases as lime extract: $K_2Cr_2O_7$ and lime extract: $SnCl_2$ while good to excellent grades to perspiration in both respective media with no colour staining.

Acknowledgement

Present work was supported by Department of Chemical Science, Hemwati Nandan Bahuguna Garhwal University (A Central University) Campus Pauri, Pauri Garhwal and Chemistry division, Forest Research Institute, Dehradun, UK, India.

References

- 1. Gaur, R. D. (2008). Traditional Dye Yielding Plants of Uttarakhand, India. *Nat. Prod. Rad.*, 7(2), 154-165.
- 2. Dayal, R. and Dobhal, P. C. (2001). Natural Dye from some Indian Plants, *Colourage*, 48(8), 33-38.
- **3.** Gulrajani, M. L. and Gupta, D. (1992). Natural Dye and their Application to Textiles. Department of Textile Technology, IIT, Delhi, 25.
- 4. Gulrajani, M. L. and Gupta, D. (1992). Introduction to Natural Dyes, Indian Institute of Technology, Delhi, 81-96.
- 5. Samanta, A. K. and Agarwal, P. (2009). Application of Natural Dyes on Textile. *Indian Journal of fibre and textile research*, 34, 384-399.
- 6. Garden, B. B. (1964). Brooklyn botanic garden. Brooklyn, N.Y. Brooklyn Botanic Garden, 1000 Washington Ave., Brooklyn, N.Y. 11225.
- 7. Trotman, E.R. (1984). Dyeing and chemical technology of textile fibres. 6th Ed. New York: John Wiley and Sons.

- 8. Singh, S. V. and Purohit, M. C. (2012). Application of ecofriendly natural dye on wool fibres using mixture of natural and chemical colour fixing agents. *Universal Journal of Environmental Research and Technology*. 2(2), 48-55.
- **9.** Bains, S., Kang, S. and Kaur, K. (2005). Dyeing of wool with *Prunuspersica* dye using mixture of colour fixing agents. *Journal of the Textile Association*, 127-131.
- **10.** Singh, S. V. and Purohit, M. C. (2014). Evaluation of colour fastness properties of natural dye isolated from *Symplocos racemosa* (Lodh) on wool fibers using mixture of botanical and chemical colour fixing agents. *Indian Journal of Fibers and Textile Research*, 39, 97-101.
- **11.** Kumaresan, *et al.* (2011). Application of Eco-friendly Natural Dye on Silk Using Mixture of Colour fixing agents. *Int. J. Chem. Res.*, 2(1), 11-14.
- **12.** Adeel, S., Ali, S., Bhatti, I. A. and Zsila, F. (2009). Dyeing of Cotton Fabric Using Pomegranate (Punica Granatum) Aqueous Extract. *Asian J. Chem*, 21(5), 3493-3499.
- **13.** Anderson, B., (1971). Creative Spinning, Weaving and Plant Dyeing. Angus and Robinson, Singapore: 24-28.
- 14. Mahangade, R. R; Varadarajan, P. V; Verma, J. K. and Bosco, H. (2009). New Dyeing Techniques for Enhancing Color Strength and Fastness Properties of Cotton Fabric Dyed with Natural Dyes. *Indian Journal of fibre and textile research*, 34(3), 279-282.
- Nishida, K. and Kabayashi, K. (1992). Dyeing properties of Natural Dyes from vegetable sources. Am Dyestuffs Rep., 81(9), 26.
- **16.** Anitha, K. and Prasad, S. N. (2007). Developing multiple natural dyes from flower parts of Gulmohur. *Current Science*, 92(12), 1681-1682.
- **17.** Siva, R. (2007). Status of natural dyes and dye yielding plants in India. *Current Science*, 92(7), 916-925.