



Frequency and Temperature dependence studies in PTh-V₂O₅ composites

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Abstract

By oxidation method, Polythiophene (PTh) has been prepared. Composites were prepared by mixing Polythiophene and V₂O₅ in different weight percentages. Their phases were confirmed by XRD and SEM studies. Dielectric properties and AC conductivity were measured over broad range of frequency and temperature. Conductivity varied with temperature in the semiconductor fashion. Dielectric constant and loss both were found to decrease with increase in frequency and increase with increase in temperature. Conductivity variation with temperature has been analysed in terms of small polaron hopping theory of Mott.

Keywords: Polythiophene, nanocomposites, conductivity, polaron hopping, activation energy.

Introduction

The conducting polymers have been of great importance as they exhibit unique optical, electrical, chemical and thermal properties. Among these polymers, polythiophene (PTh) received attention due to its high conductivity and thermal stability¹⁻⁴. The conductivity of these materials can be tuned by doping. Dopant anion plays an important role in polymerization^{5,6}. Polymer nanocomposites are considered to be hybrid. Nano composites of polymers are used in drug delivery, conductive paints, batteries etc⁷⁻¹³. Vanadium oxides are polyfunctional materials and show metal semiconductor transition. V₂O₅ is used as an active electrode in a rechargeable lithium battery, electrochromic devices, catalysts etc¹⁴⁻¹⁹. The conductivity of PTh-ZnO composites was measured and found to be of the order of 10⁻⁴ Ω/m⁷. The room temperature conductivity of polypyrrole-TeO₂ and PTh-TeO₂ composites has been reported to be 1 × 10⁻⁵ Ωcm⁻¹ and 2 × 10⁻² Ωcm⁻¹ respectively²⁰. In these composites, conductivity has been observed to have increased by 10³ to 10⁴ orders of magnitude compared to their pure PPy and PTh. PPy-V₂O₅ composites of different wt% in the range from 10 to 50 wt% were prepared by chemical oxidative method. The room temperature conductivity of these composites at 100 KHz revealed that the addition of vanadium oxide nanoparticle results in the decrease in conductivity up to 10% V₂O₅ and remains constant for higher amounts of V₂O₅. The AC conductivity of pure PPy increased with frequency and PPy-V₂O₅ composite showed constant up to the frequency of 10⁵ Hz and then increased steeply¹⁶. Polyaniline-V₂O₅ composites showed constant conductivity up to 10⁵ Hz^{17, 18}. DC conductivity of polyaniline-V₂O₅ composites of different wt% of V₂O₅ was found to change from 10⁻⁷ to 10⁻⁹ Ωcm⁻¹, attaining a maximum value for 30% V₂O₅¹⁵. Here, we present the result of dielectric results for PTh-V₂O₅ nanocomposites. AC conductivity has been determined using dielectric data. Frequency and temperature dependence of both dielectric data and conductivity

data has been thoroughly analyzed.

Material and Methods

PTh was prepared at 323K using AR grade Thiophene, Ferric chloride, Methanol and Chloroform. Homogenous aqueous solution of thiophene was prepared. Chloroform and ferric chloride solutions were mixed drop by drop to the PTh solution. The mixture magnetically stirred for 24 hours and filtered. The precipitate so formed was washed with chloroform and then with methanol. In this process, the precipitate changed its colour to brown indicating the formation of Polythiophene. The powder was dried up and subsequently grinded^{20,21}. The PTh-V₂O₅ composites were prepared by mixing Polythiophene and analytical grade V₂O₅ in different wt% defined as (PTh) 100-x (V₂O₅)x, where x = 5%, 10% and 15% labelled as PTh-VO1, PTh-VO2 and PTh-VO3 respectively. Powders were subjected to XRD and SEM studies. From these studies, it is confirmed that the grains in these composites are of nano size²². Powders of the composites were pressed into pellets. Capacitance, C, and dissipation factor, tanδ, were measured as a function of frequency and temperature in the range from 50Hz to 3MHz and 300K to 423K respectively. These measurements were carried out in a precision impedance analyser (Wayne Kerr make Model No. 6500B). Temperature was sensed using Chromel-Alumel thermo-couple with the accuracy of ± 1K.

Results and Discussion

Dielectric properties: Using the measured capacitance and dissipation factor the dielectric parameters were determined for all the composites, using the expressions given in reference²³. Figure-1 shows the variation of dielectric constant, ε' with frequency for all the three composites. From this figure, we note that, ε' decreases gradually with frequency up to 70 KHz and becomes constant for higher frequencies. Dielectric loss factor, ε'' variation with frequency for all the three composites is

plotted in figure-2. ϵ'' varied with frequency in the same fashion as that of ϵ' . Temperature variation of ϵ' for PTh-VO1 is shown in figure-3. In this figure, we see that ϵ' increases with temperature. Similar nature of variation of ϵ' with temperature has been observed for the remaining composites. The change in ϵ'' with temperature for PTh-VO1 is shown in figure-4. Similar behaviour of ϵ'' with temperature has been observed for the remaining composites.

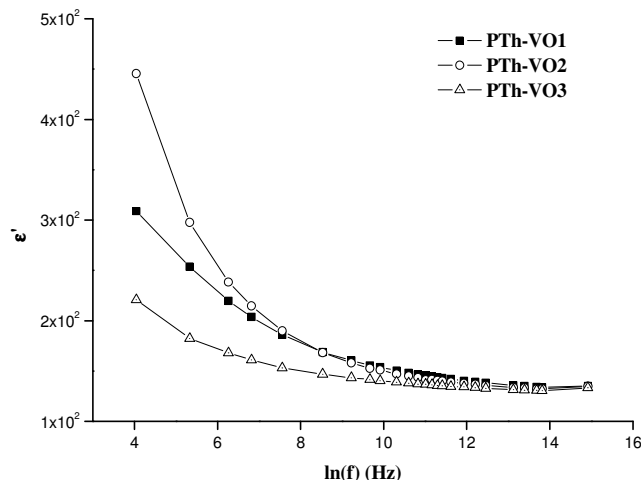


Figure-1
Dielectric constant, ϵ' versus $\ln(f)$ for PTh-VO nanocomposites at the temperature of 323K

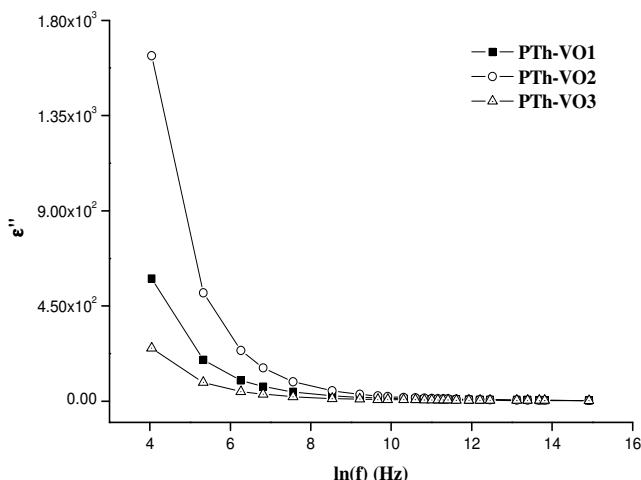


Figure-2
Dielectric loss, ϵ'' versus $\ln(f)$ for PTh-VO nanocomposites at the temperature of 323K

Electrical Conductivity: Conductivity, σ , has been determined using dielectric data using the following equation ²⁴.

$$\sigma_{ac} = \epsilon'' \omega \epsilon_0 \quad (1)$$

Where: ϵ_0 is free space permittivity which is equal to $8.85 \times 10^{-12} \text{ Fm}^{-1}$. Conductivity variation with temperature for different

frequencies for the composite PTh-VO2 is shown in figure-5.

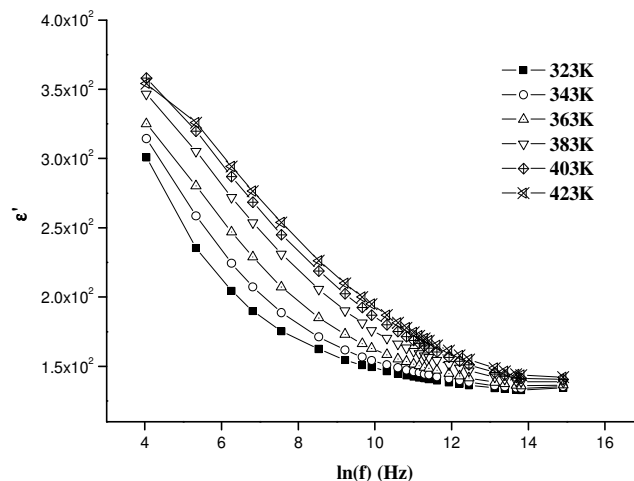


Figure-3
Dielectric constant, ϵ' versus $\ln(f)$ for PTh-VO1 nanocomposites for different Temperatures

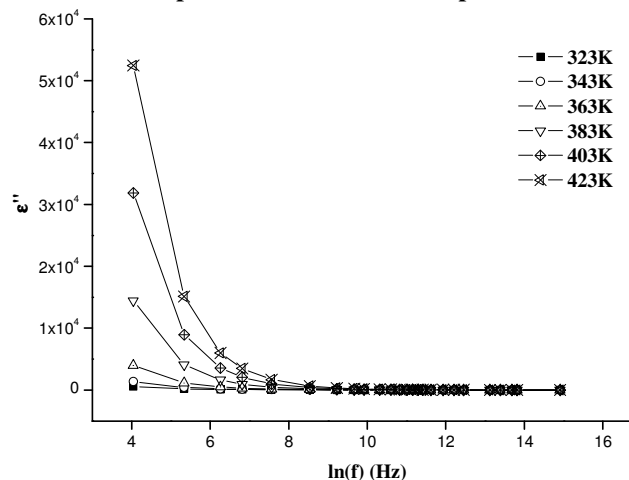


Figure-4
Dielectric loss, ϵ'' versus $\ln(f)$ for PTh-VO1 nanocomposites for different temperatures

It can be seen in figure-5 that conductivity increases with increasing temperature indicating semiconducting type of behaviour. It also increased with increasing frequency. Similar results have been reported for polyaniline doped with silver nanoparticles, polyaniline doped with cobalt and polyaniline doped with nickel oxide²⁵⁻²⁷. All the present composites behaved in the same way. Variation of conductivity of all the present composites was found to be within the same order of magnitude i.e., $10^{-4} (\Omega^{-1}\text{m}^{-1})$.

Conductivity variation with V_2O_5 content for two different frequencies at temperature of 403K is shown in figure-6. From figure-6 it is clear that conductivity decreased with increasing V_2O_5 content.

The conductivity data as a function of temperature has been fit

to the Mott Small Polaron Hopping (SPH) theory. This theory gave the conductivity expression as²⁸,

$$\sigma = \frac{\sigma_0}{T} \exp\left\{-\frac{E_a}{K_B T}\right\} \quad (2)$$

Where E_a the activation energy for small polaron hopping.

The plots of $\ln(\sigma T)$ versus $(1/T)$ were made as per equation-2 for the composite PTh-VO₂ and shown in figure-7. The linear lines were fit to the data in the high temperature region where the data appeared linear. Activation energy, E_a for ac conductivity was calculated using the slopes of the fit linear lines.

Activation energy, E_a versus V₂O₅ content determined for the present composites for different frequencies are plotted in figure-8.

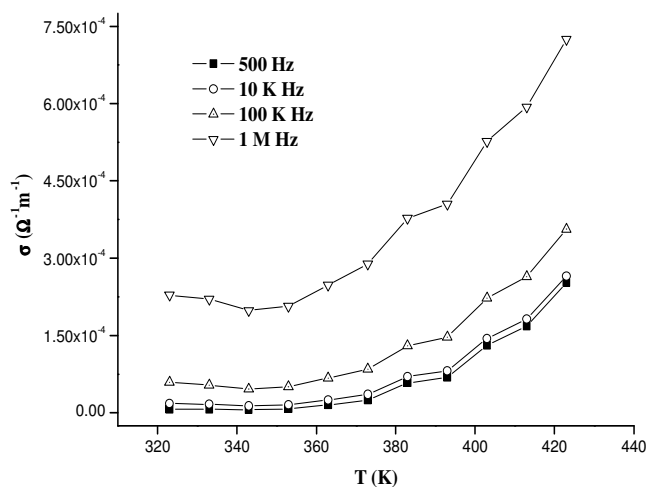


Figure-5

Temperature dependence of electrical conductivity of PTh-VO₂ composite nanoparticles at different frequencies

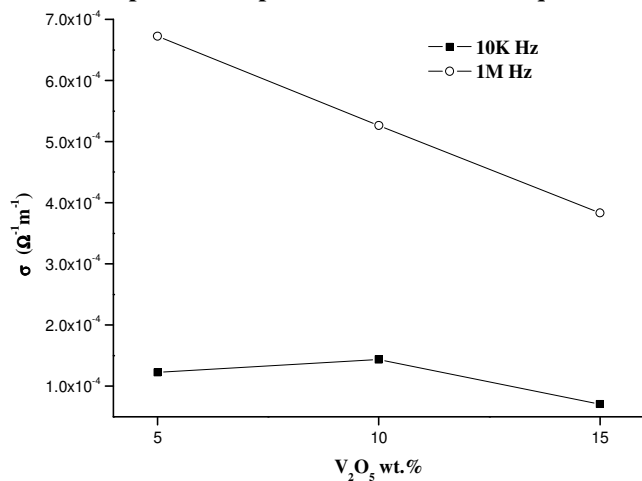


Figure-6

Conductivity versus wt % of V₂O₅ in PTh-VO nanocomposites for two different frequencies at T=403K

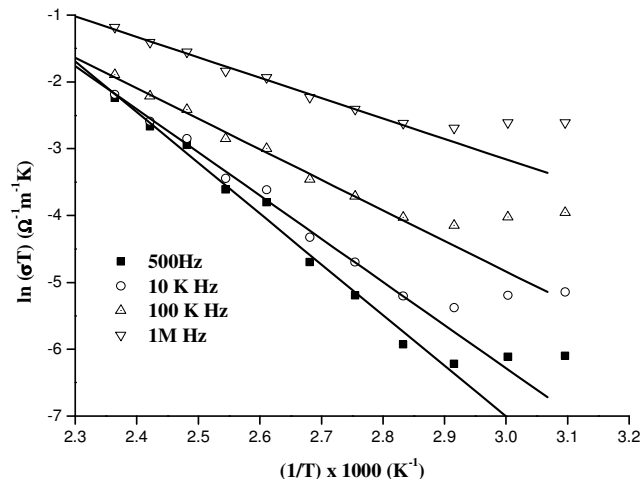


Figure-7

Plots of $\ln(\sigma T)$ versus $(1/T)$ for PTh-VO₂ composite for four different frequencies. Solid lines are linear fits as per Mott's SPH model

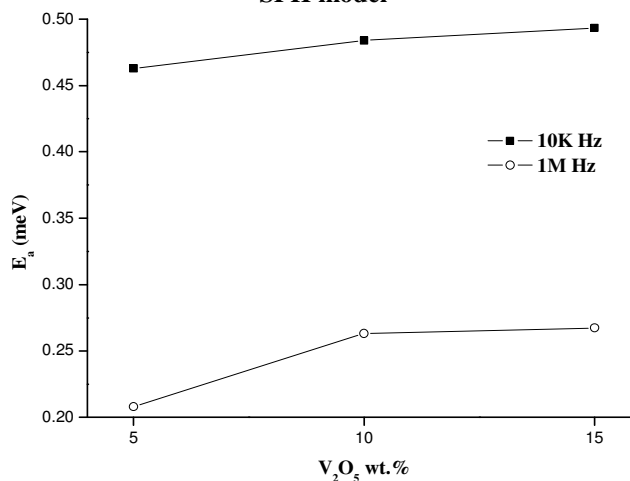


Figure-8

Activation energy versus wt. % of V₂O₅ for PTh-VO nanocomposites at two different frequencies

From the figure-8 one can note that E_a increases with increase of V₂O₅ content and decreased with increase of frequency. Increase in E_a with increase in V₂O₅ concentration may be that addition of V₂O₅ content to the PTh network contributes more to the scattering of polarons. Similar results have been reported for polyaniline doped with V₂O₅, polyanilinedoped with CeO₂ and polypyrrole doped with Ag^{17,29,30}.

Conclusion

Polythophene has been synthesised at 323K by chemical method. PTh-V₂O₅ composites were prepared by mixing Polythiophene and V₂O₅ in different weight percentages. The changes in dielectric properties with temperature and frequency were measured over broad ranges. Dielectric constant decreased with increase in frequency and increased with temperature. Dielectric loss decreased with increase in frequency and

increased with temperature. Conductivity increased with increase of both temperature and frequency. By employing Mott's Small Polaron Hopping Model's conductivity expression, activation energy for ac conductivity has been obtained. Activation energy was found to be decreased with increase in frequency and increased with V₂O₅ content and it may be attributed to increase in the scattering rate of polarons with increase in V₂O₅ content.

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